



atom the smallest unit of matter that retains all the chemical properties of an element isotopes are forms of a chemical

an isotope is an atom that has the same number of protons and a different number of neutrons the atom is still considered the same element still contains the same number of protons as before but

sep 22 2024 definition atoms of an element which have the same atomic number but different mass numbers are called isotopes of that element these isotopes are variants of the same element in

the meaning of isotope is any of two or more species of atoms of a chemical element with the same atomic number and nearly identical chemical behavior but with differing atomic mass or mass

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